



The enhanced hydrogen storage combination of 6Mg(NH₂)₂-9LiH-LiBH₄ observed in (Mg-B-N-H)-doped Mg(NH₂)₂-2LiH system

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Doping Mg(NH₂)₂-2LiH by 0.1mol Mg₂(BH₄)₂(NH₂)₂ compound exhibits enhanced hydrogen de/re-hydrogenation performance. The peak width in temperature-programmed desorption (TPD) profile for the Mg(NH₂)₂-2LiH-0.1Mg₂(BH₄)₂(NH₂)₂ was remarkably shrunk by 60°C from that of pristine Mg(NH₂)₂-2LiH, and the peak temperature was lowered by 12°C from the latter. Its isothermal dehydrogenation rate was greatly improved by five times from the latter at 200°C. XRD, FTIR and NMR analyses revealed that a series of reactions occurred in the dehydrogenation of the composite. The prior interaction between LiH and Mg-B-N-H yielded intermediate LiBH₄, Mg₃N₂ and Li₂Mg(NH)₂, which can be described by a merged reaction of $6\text{LiH} + 2\text{Mg}(\text{BH}_4)_2\text{-Mg}(\text{NH}_2)_2 \rightarrow 4\text{LiBH}_4 + \text{Mg}_3\text{N}_2 + \text{Li}_2\text{Mg}(\text{NH})_2 + 6\text{H}_2 \uparrow$. The LiBH₄ and Mg₃N₂ phases can be regarded as the meditation product between LiH with Mg(BH₄)₂, and the rapidly subsequent reaction product between intermediate MgH₂ with Mg(NH₂)₂, respectively. Li₂Mg(NH)₂ was generated from the interaction between LiH and Mg(NH₂)₂. Only a few Mg(NH₂)₂ joined in the reaction with LiH/MgH₂ but most was remained until the temperature was lifted to 180°C.

The formed LiBH₄ subsequently reacted with Mg(NH₂)₂ and LiH in molar ratio of 1:6:9 to form Li₂Mg₂(NH)₃ and Li₄BN₃H₁₀ phases. The observed 6Mg(NH₂)₂-9LiH-LiBH₄ combination dominated the hydrogen release and soak in the Mg(NH₂)₂-2LiH-0.1Mg₂(BH₄)₂(NH₂)₂ system. Such combination can be reflected by $6\text{Mg}(\text{NH}_2)_2 + 9\text{LiH} + \text{LiBH}_4 \rightarrow \text{Li}_4(\text{BH}_4)(\text{NH}_2)_3 + 3\text{Li}_2\text{Mg}_2(\text{NH})_3 + 9\text{H}_2 \uparrow$, which is thermodynamically reversible. It can be found that the thermal stability of the reversible system 2Mg(NH₂)₂-3LiH / Li₂Mg₂(NH)₃-LiNH₂ was effectively altered by doping 1/3 mol LiBH₄. The kinetics of the system was also enhanced by additive LiBH₄.

Keywords: Hydrogen storage; Mg-B-N-H compounds; Kinetics; Doped Mg(NH₂)₂-2LiH system